

IN THE SPECIFICATION:

Please insert the following paragraph at the beginning of the specification.

This application is a 371 of international application PCT/JP2004/004182, which claims priority based on Japanese patent application Nos. 2003-101226 filed April 4, 2003, and 2003-186632 filed June 30, 2003, which are incorporated herein by reference.

Please replace the paragraph beginning on page 10, line 9, with the following rewritten paragraph:

In this invention, the solvent containing an ester structure is preferably used, and a solvent containing a lactone structure is more preferable. The most preferable solvent is γ -butyrolactone. The boiling point used in this invention is the boiling point under one atmospheric pressure, i.e., the pressure of ~~$1.013 \times 10^5 \text{ N/m}^2$~~
 $1.013 \times 10^5 \text{ N/m}^2$. Although the measurement of boiling point can be done by a well-known technique and it is not especially limited, it can be measured by using, for example, the boiling point meter of Swietoslawski.

Please replace the paragraph beginning on page 26, line 22,

with the following rewritten paragraph:

(2) Supposing the weight of substrate as W1, the weight of [[glass]] substrate and the dielectric composition as W2, the density of the dielectric composition as D, and the volume as V, the dielectric composition, $D = (W2-W1)/V$.

Please replace the paragraph beginning on page 37, line 11, with the following rewritten paragraph:

Except that the solvent was propylene glycol monomethyl monomethylether acetate, a paste composition D-3 was prepared in the same way as that of the paste composition C-2. The boiling point of propylene glycol monomethyl acetate is 146°C. Then, according to the method of Example 1, a high dielectric constant composition was prepared and the result of evaluation of its dielectric characteristics is shown in Table 4. The relative dielectric constant was 46, the dielectric loss tangent was 4.7%, and the capacitance per area was 2.7 nF/cm², and was inferior in the electrical property. The porosity was 35 volume%.

Please replace the paragraph beginning on page 49, line 9, with the following rewritten paragraph:

A barium titanate filler (BT-05 of SAKAI CHEMICAL INDUSTRY

Co., Ltd., mean particle diameter: 0.5 μm) 6067 weight parts, a barium strontium titanate filler (HPS-2000 of TPL. Inc., mean particle diameter: 0.045 μm) 1613 weight parts, γ -butyrolactone 1523 weight parts and a dispersant (a copolymer having an acid group with a phosphoric-ester skeleton: BYK-W9010 of BYK-Chemie Japan KK) 77 weight parts were mixed and dispersed under ice-cooling for 1 hour using a homogenizer, and a dispersion liquid X-7 was obtained.

Please replace the paragraph beginning on page 52, line 14, with the following rewritten paragraph:

The epoxy resin (~~"Phenolite"~~ EPPN-502H of NIPPON KAYAKU CO., LTD.) 400 weight parts, a phenol novolak resin (TD-2131 of DAINIPPON INK AND CHEMICALS, Inc.) 400 weight parts and γ -butyrolactone 1000 weight parts were mixed and the resin solution Y-1 was obtained.

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Table 1 beginning on page 63 has been amended as follows:

Paste composition							Dielectric Characteristics (1 MHz)				Film Characteristic
Example	Inorganic filler	Resin	Curing agent	Solvent	Additive agent	Content of the inorganic filler in the solid content (wt%)	Thickness (μm)	Releative dielectric constant	Capacitance (nF/cm ²)	Dielectric loss tangent	Porosity (volume %).
1	Barium Titanate	Epoxy resin NIPPON SAKAI CHEMIC AL BT-05	Novolak resin EPPN50 2H	Y-butyrolactone resin DAINIP PON INK TD2131	triphenylphosphine	94	10	10	82	7.3	2.8

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2	Barium Titana- te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola- k resin DAINIP PON INK TD2131	γ - butyro- lacton e	triphe- nylpho- sphine	94 15 15 73 4.3 3.4 12	
3	Barium Titana- te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola- k resin DAINIP PON INK TD2131	γ - butyro- lacton e	triphe- nylpho- sphine	94 20 10 65 5.8 3.0 14	
4	Barium Titana- te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola- k resin DAINIP PON INK TD2131	γ - butyro- lacton e	triphe- nylpho- sphine	94 25 8 58 6.4 3.2 20	

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5	Barium Titana- te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola- k resin DAINIP PON INK TD2131	γ - butyro- lacton- e	triphe- nylpho- sphine	94	10	8	102	11.3	3.6	6
6	Barium Titana- te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola- k resin DAINIP PON INK TD2131	γ - butyro- lacton- e	triphe- nylpho- sphine	94	15	10	95	8.4	3.1	7
7	Barium Titana- te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola- k resin DAINIP PON INK TD2131	N- methyl 1-2- pyrrol- idone	triphe- nylpho- sphine	94	15	10	58	5.3	4.6	2.6

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8	Barium Titana- te	Epoxy resin NIPPON KAYAKU	Phenol Novola- k resin	Ethyle ne glycol acetat DAINIP PON	triphe- nylpho- sphine acetat diacet- ate	94 15 10	64 5.7	4.8	21
	SAKAI CHEMIC AL BT- 05	EPPN50 2H		INK TD2131					

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Table 8 on page 72 has been amended as follows:

Example	Paste composition					Resin solution	Inorganic filler/resin ratio	Stability of dispersion on liquid
	Dispersed on liquid	Inorganic filler	Inorganic filler composition	Mean particle diameter (μm)	Mean particle diameter (μm)			
X-2	Barium Titanate	0.5	Barium Titanate	0.060	8.3	Y-1	79/21	Stability
X-7	Barium Titanate	0.5	Strontium Titanate	0.045	11.1	Y-1	79/21	Stability
X-8	Barium Titanate	0.5	Titanium Oxide	0.2	2.5	Y-1	[81/29] 81/19	Instability slightly (cohesion)
X-9	Lead type filler	0.9	Barium Titanate	0.059	15.3	Y-1	86/14	Stability

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			Y-1	0/100	-
Comparat ive example 4	-				
Comparat ive example 5	X-10	Barium Titanate	0.5	-	Y-1 79/21 Stabilit y
Comparat ive example 6	X-11	Barium Titanate	7	Barium Titanate 0.5	14 79/21 Instabil ity (filler sediment ation)
Comparat ive example 7	X-12	Barium Titanate	40	Barium Titanate 2.1	19 79/21 Instabil ity (filler sediment ation)
Comparat ive example 8	X-13	Barium Titanate	20	Barium Titanate 2.1	9.5 79/21 Instabil ity (filler sediment ation)
Comparat ive example 9	-	Barium Titanate	0.059	Strontiu m Titanate 0.045	1.3 79/21 Instabil ity (cohesio n)